## Sesquiterpenes from *Baeckea frutescens*

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The aerial parts of *Baeckea frutescens* yielded three sesquiterpenes (1-3) belonging to the humulane, caryophyllane, and clovane classes, including the new sesquiterpene (–)-clovane-2,9-diol (3). Circumstantial evidence is presented that these classes are biogenetically related.

*Baeckea frutescens* L. (syn. *Baeckea chinensis* Gaertner; Myrtaceae) is widespread in Hong Kong and is used in traditional Chinese medicine for treating rheumatism and snake bites.<sup>1</sup> There has been one previous phytochemical study of *B. frutescens*, which reported three phloroglucinols.<sup>2</sup>

Column chromatography followed by HPLC of a CH<sub>2</sub>Cl<sub>2</sub> extract of the aerial parts of *B. frutescens* yielded three sesquiterpenes (1-3), which were identified largely on the basis of 2D-NMR spectroscopy. Analysis of the results of the PFG-HSQC, the PFG-HMBC, and the <sup>1</sup>H<sup>-1</sup>H COSY (proton–proton couplings) NMR spectra (500 MHz) established the structure of 1 and also allowed determination of chemical shifts for all carbons and protons in the molecule. Compound 1 has been isolated previously as a constituent of Zingiber zerumbet and named humulene epoxide II;<sup>3-6</sup> our spectra agreed well with <sup>13</sup>C- and partial <sup>1</sup>H-NMR data reported for humulene epoxide II in the literature.<sup>3,7,8</sup> Full <sup>13</sup>C- and <sup>1</sup>H-NMR assignments for **1** are presented in Table 1 for the first time. The optical rotation of humulene epoxide II from nature is variously reported as  $-25^{\circ}$ ,  $-31^{\circ}$ ,  $-10^{\circ}$ and -43°,<sup>4,5,8,9</sup> whereas 1 from *Baeckea frutescens* had zero optical rotation and must therefore be a racemic mixture of trans epoxides.



The <sup>13</sup>C-NMR signals (125 MHz) for the C-12- and C-13 methyl groups in 1 showed strong line broadening relative to other peaks. This broadening was progressively reduced as the temperature was raised, both resonances attaining a normal line-shape at 60 °C in  $C_6H_6$ - $d_6$ . Such behavior would be consistent with a degree of conformational rigidity in the portion of the 11-membered ring next to the gem-dimethyl group, relative to the rest of the molecule, which is overcome at higher temperatures. In support of this, correlations in the NOESY spectrum clearly demonstrated that there is a preferred conformation for the two double bonds flanking the C-12 and C-13 methyl groups (as shown in Scheme 1) at room temperature (NOESY gave no clear indication of a conformational preference for the other half of the ring). Observation of this conformational preference is interesting because the two double bonds are almost parallel, and the alkene protons at the C-2 and C-10 positions project towards different faces of the molecule. This same conformation has been proposed to account for rearrangements of humulene epoxide II (1) in superacid and has been suggested to be the most stable conformer from molecular modeling studies.<sup>10</sup> Our own molecular modeling calculations (using the MM2\* force field of Macromodel<sup>11</sup>) failed to show a clear conformational preference for 1: the minimized energies for the five lowest energy conformers were within 10 kJ mol<sup>-1</sup> of each other. Conformations calculated for the second and third lowest energy states incorporated the partial structures shown for conformers **a** and **b** in Scheme 1.

The observed conformation for **1** would naturally yield *trans*-caryophyllanes (possibly via an intramolecular ene reaction), which are often found in association with humulanes. In support of this possibility, we have isolated two other sesquiterpenes (**2** and **3**) from the same extract whose structures are consistent with such cyclization of  $(\pm)$ -humulene epoxide II.

Compound **2** was identified as the known compound caryophyllene- $4\beta$ , $5\alpha$ -epoxide (1R,4R,5R,9S) from its NMR spectra (300 MHz) and optical rotation.<sup>12–16</sup> Our assignments of the proton and carbon resonances for **2** agreed with those reported in the literature.<sup>17</sup> Compound **2** can clearly be formed from **1** by direct cyclization of the 6R,7R form of **1**, as postulated above (Scheme 2).

Compound 3 was identified as the novel compound (-)-clovane-2,9-diol (1R,2R,5R,8S,9S) in the following manner. Accurate mass spectrometry of 3 demonstrated the molecular formula C<sub>15</sub>H<sub>26</sub>O<sub>2</sub>. IR spectroscopy showed a strong band at 3383 cm<sup>-1</sup> indicative of hydroxyl group functionality(ies). The <sup>13</sup>C/DEPT-NMR spectrum (125 MHz) indicated 15 carbons with 24 directly attached protons and analysis of the chemical shifts suggested the presence of two oxygenated carbons ( $\delta_{\rm C}$  80.9 CH, 75.1 CH) and the absence of double bonds. Consequently, **3** was established as a tricyclic compound with two hydroxyl groups. The complete structure for **3** and assignments for all proton and carbon resonances were deduced from analysis of 2D-NMR spectra (500 MHz) (high-resolution spectra were required because of the overlap of several proton chemical shifts), and the relative stereochemistry of 3 was deduced from correlations observed in the NOESY spectrum (Table 2). In fact, both the <sup>1</sup>H- and <sup>13</sup>C-NMR spectra for **3** agreed well with those reported for clovane-2,9-diol (1S, 2S, 5S, 8R, 9R).<sup>12,13,18</sup> The optical rotation for **3** from *B. frutescens* ( $[\alpha]_D$  –3.5°) was of opposite sign to that

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position	$\delta_{ m C}$	multiplicity <sup>a</sup>	$\delta_{ m H}$	HMBC correlation from $\delta_{\rm H}$ to $\delta_{\rm C}$	$^{1}H^{-1}H COSY$ correlation ( $\delta$ )
1	40.2	$CH_2$	1.99	131.9, 125.7, 36.5, 29.0	4.99, 1.87
			1.87	131.9, 125.7	4.99, 1.99
2	125.7	СН	4.99	36.6, 15.1	1.99, 1.87
3	131.9	С			
4	36.6	$CH_2$	2.20	131.9, 125.7, 24.8	2.10, 2.14
			2.10	131.9, 125.7, 24.8	2.20
5	24.8	$CH_2$	2.14	62.0	2.20, 1.37
			1.37	63.2, 36.6	2.52, 2.14
6	62.0	СН	2.52	24.8	1.37
7	63.2	С			
8	42.5	$CH_2$	2.57	143.1, 122.1, 63.2, 17.2	5.29, 1.65
			1.65	143.1, 122.1, 63.2, 17.2	5.92, 2.57
9	122.1	СН	5.29	143.1, 63.2, 42.6, 36.5	5.16, 2.57, 1.65
10	143.1	СН	5.16	122.1, 42.6, 36.5, 29.0, 25.6	5.29
11	36.5	С			
12	29.0	$CH_3$	1.09	143.1, 40.2, 36.5, 25.6	
13	25.6	$CH_3$	1.11	143.1, 40.2, 36.5, 29.0	
14	17.2	$CH_3$	1.31	63.2, 42.6	
15	15.1	$CH_3$	1.57	131.9, 125.7, 36.6	

<sup>a</sup> Values were determined from the DEPT spectrum.

**Table 2.** NMR Data for Compound 3

Table 1. NMR Data for Compound 1

position	$\delta_{\mathrm{C}}$	multiplicity <sup>.a</sup>	$\delta_{\mathrm{H}}$	HMBC correlation from $\delta_{\rm H}$ to $\delta_{\rm C}$	$^{1}H^{-1}H COSY$ correlation ( $\delta$ )	NOESY correlation $(\delta)$
1	44.2	С				
2	80.9	СН	3.79	47.6, 35.5, 26.4	1.72, 1.51	1.72, 1.56, 1.51, 0.91, 0.86
3	47.6	$CH_2$	1.72	80.9, 50.5, 44.2, 37.2, 25.4	3.79, 1.51	3.79, 1.51, 0.86
			1.51	80.9, 44.2, 37.2, 31.5, 25.4	3.79, 1.72	3.79, 1.72, 1.03
4	37.2	С				
5	50.5	СН	1.43	44.2, 37.2, 35.5, 31.5, 26.4, 25.4, 20.7		2.00, 1.03
6	20.7	$CH_2$	1.42	50.5, 34.7, 33.1	1.32, 1.12	1.12
			1.32	50.5, 37.2, 33.1	1.42, 1.39, 1.12	0.86
7	33.1	$CH_2$	1.39	75.1, 50.5, 34.7, 28.3, 20.7	1.32, 1.12	0.96
			1.12	75.1, 34.7, 28.3, 20.7	1.42, 1.39, 1.32	3.33, 1.42
8	34.7	С				
9	75.1	СН	3.33	35.5, 26.4	2.00, 1.64	2.00, 1.64, 1.12, 0.96
10	26.0	$CH_2$	2.00	26.4	3.33, 1.68, 1.64, 1.08	3.33, 1.64, 1.43
			1.64	75.1, 44.2, 34.7	3.33, 2.00, 1.68, 1.08	3.33, 2.00
11	26.4	$CH_2$	1.68	50.5, 44.2, 35.5, 26.0	2.00, 1.64, 1.08	1.08
			1.08	35.5	2.00, 1.68, 1.64	1.68
12	35.5	$CH_2$	1.56	75.1, 50.5, 44.2, 34.7, 33.1, 28.3, 26.4	0.91	3.79, 0.91
			0.91	80.9, 75.1, 44.2, 34.7, 28.3, 26.4	1.56	3.79, 1.56
13	25.4	$CH_3$	0.86	50.5, 47.6, 37.2, 31.5		3.79, 1.72, 1.32, 1.03
14	31.5	$CH_3$	1.03	50.5, 47.6, 37.2, 25.4		1.51, 1.43, 0.86
15	28.3	CH <sub>3</sub>	0.96	75.1, 35.5, 34.7, 33.1		3.33, 1.39

<sup>a</sup> Values were determined from the DEPT spectrum.

**Scheme 1.** Preferred Conformations for **1** (**a** and/or **b**) As Deduced from NOESY Correlations (Indicated by Double-Headed Arrows)



reported in the literature (variously given as +3.2° or +5° from both natural and synthetic sources, <sup>13,19</sup> (+)clovane-2,9-diol (1*S*,2*S*,5*S*,8*R*,9*R*) prepared by ourselves from **2** (commercially available) according to a literature procedure<sup>19</sup> gave  $[\alpha]_D$  +3.8° (*c* 4.5, CHCl<sub>3</sub>). It thus seems likely that **3** is the mirror image of the compound reported in the literature: we propose the name (-)-clovane-2,9-diol, associated with the absolute stereo-chemistry (1*R*,2*R*,5*R*,8*S*,9*S*).

Compound **3** can be derived from the 6S,7S form of **1** by cyclization to caryophyllene oxide

Scheme 2. Proposed Conversion of 1 (Racemic) into 2 and 3



(1.5,4.5,5.5,9.R) followed by subsequent rearrangement to the clovane skeleton (Scheme 2). (+)-Clovane-2,9diol (1.5,2.5,5.5,8.R,9.R) has been reported to arise from **2** both naturally and synthetically by just such a rearrangement.<sup>13, 19</sup>

## **Experimental Section**

General Experimental Procedures. FTIR spectra were recorded in  $CCl_4$  on a Shimadzu FTIR-8201 PC

instrument. All NMR experiments were performed on a Bruker DPX 300 or a DRX 500 instrument with CDCl<sub>3</sub> as solvent. PFG–HSQC and PFG–HMBC spectra were normally recorded with 2048 data points in F<sub>2</sub> and 128 data points in F<sub>1</sub>, while high-resolution experiments had 8192 data points in F<sub>2</sub> and 1024 data points in F<sub>1</sub>. MS were recorded in EI mode (70 eV) on a Finnigan-MAT 95 MS spectrometer. Column chromatography was made on Si gel Merck (60–200  $\mu$ m). TLC plates were developed using *p*-anisaldehyde. HPLC separations were performed using a PREP-SIL 20 mm × 25 cm column, flow rate 8 mL/min, with refractive index detection.

**Plant Material.** *Baeckea frutescens* was collected in November 1995, during flowering at Plover Cove Country Park, New Territories, Hong Kong. A voucher specimen is deposited in the University of Hong Kong Herbarium (GDBROWN 96/1).

**Extraction and Isolation.** The aerial parts of *B. frutescens* (577 g) were ground to a fine powder under liquid N<sub>2</sub> and immediately extracted with CH<sub>2</sub>Cl<sub>2</sub> in a Soxhlet apparatus (8 h). The organic extract was then dried and evaporated under vacuum to yield a dark green oil (8.85 g; 1.5% w/w), which was separated chromatographically as follows: **1** (15 mg) by column chromatography ( $R_f$  0.25 in 6% EtOAc/hexane; staining violet) followed by HPLC ( $t_R$  15.9 min in 6% EtOAc/hexane; staining yellow) followed by HPLC ( $t_R$  16.4 min in 6% EtOAc/hexane); **3** (7 mg) by column chromatography ( $R_f$  0.12 in 50% EtOAc/hexane; staining pink) followed by HPLC ( $t_R$  23.9 min in 50% EtOAc/hexane).

(±)-Humulene epoxide II (1): obtained as a colorless oil;  $[\alpha]_D 0.0^\circ$  (*c* 0.64, CHCl<sub>3</sub>, ); IR  $\nu$  max 2961, 2930, 2870, 1690, 1460, 1448, 1387, 1074 cm<sup>-1</sup>; <sup>1</sup>H NMR (500 MHz)  $\delta$  5.29 (1H, ddd, *J* = 15.8, 10.2, 5.1 Hz, H-9), 5.16 (1H, d, *J* = 15.8 Hz, H-10), 4.99 (1H, dd, *J* = 9.5, 5.2 Hz, H-2), 2.57 (1H, dd, *J* = 12.2, 5.1 Hz, H-8), 2.52 (1H, dd, *J* = 10.2, 3.9 Hz, H-6), 1.99 (1H, dd, *J* = 13.7, 9.2 Hz, H-1), 1.87 (1H, dd, *J* = 13.7, 5.2 Hz, H-1), 1.65 (1H, dd, *J* = 12.2, 10.2 Hz, H-8), 1.57 (3H, s, H-15), 1.31 (3H, s, H-14), 1.11 (3H, s, H-13); 1.09 (3H, s, H-12); <sup>13</sup>C-NMR data, see Table 1; EIMS *m*/*z* [M<sup>+</sup>] 220.1822 ( $\Delta$  +0.5 mmu for C<sub>15</sub>H<sub>24</sub>O) (2), 138 (36), 133 (30), 111 (30), 109 (100), 107 (57).

**Caryophyllene**  $4\beta$ , $5\alpha$ -**Epoxide (2).** Data were in agreement with published values.<sup>12–18</sup>

(-)-Clovane-2,9-diol (3): obtained as a white solid; mp 147–149 °C;  $[\alpha]_D$  –3.5° (*c* 0.16, CHCl<sub>3</sub>); IR  $\nu$  max 3383, 2952, 2930, 2868, 1463, 1074 cm<sup>-1</sup>; <sup>1</sup>H NMR (500 MHz)  $\delta$  3.79 (1H, dd, J = 10.0, 6.0 Hz, H-2), 3.33 (1H, s, H-9), 1.03 (3H, d, J = 2.0 Hz, H-14), 0.96 (3H, s, H-15), 0.86 (3H, s, H-13); <sup>13</sup>C-NMR data, see Table 2; EIMS m/z [M<sup>+</sup>] 238.1929 ( $\Delta$  +0.3 mmu for C<sub>15</sub>H<sub>26</sub>O<sub>2</sub>) (25), 220 (57), 182 (49), 179 (42), 164 (100), 161 (32), 123 (18).

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